

Alexander K. Tucker-Schwartz and Robin L. Garrell
Department of Chemistry and Biochemistry & California NanoSystems Institute
University of California, Los Angeles

Introduction

Polymer sphere and shell applications:

- Controlled drug delivery
- Inertial confinement fusion targets
- Photonic crystals
- Catalytic supports
- Sensors

Goals

Use droplet-based and channel microfluidic devices to form polymer spheres or shells with controlled, size, shape and concentration.

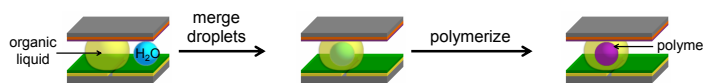
Current polymer sphere and shell formation problems:

- Polydispersity
- Aggregation
- Poor concentricity
- Lack of control over shell wall uniformity

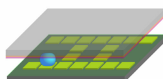
Droplet-Based Microfluidics Approach

Shell Formation

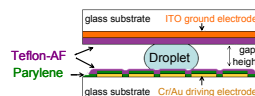
- Encapsulate water droplet within an immiscible organic droplet between two hydrophobic surfaces
- Polymerize monomer at the interface



3-D Device Schematic



2-D Device Schematic



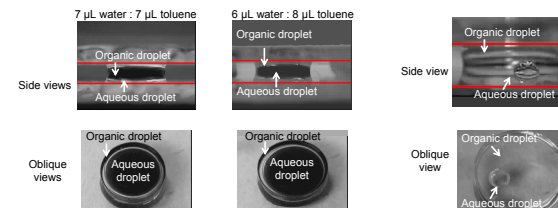
Droplet Driving Mechanisms



Droplet Entrapment and Polyaniline Beads

Aqueous droplet inserted into organic droplet between two hydrophobic surfaces

- Dyed water droplet encapsulated in toluene between Teflon coated plates
- Smaller embedded water droplet is spherical



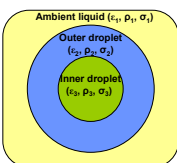
Interfacial polymerization of aniline by ammonium persulfate (APS) leads to hollow oblate polymer shells.



Double Emulsion Droplet Levitation and Centering

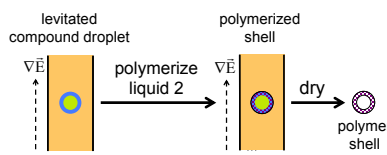
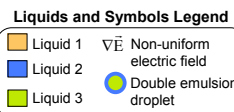
Project Objectives

- Use dielectrophoresis (DEP) to levitate double emulsion (droplet in a droplet)
- Use DEP to center inner droplet (The concentric outer droplet can be polymerized to form a shell)



- Levitation counteracts gravity, lifting sessile droplet into ambient liquid
- Suspended droplet becomes spherical to minimize liquid-liquid interfacial area
- DEP centers inner droplet forming concentric outer liquid shell

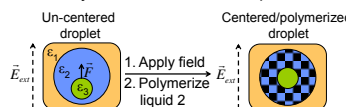
Droplet Levitation



DEP Centering

Centering approach

- Apply uniform electric field to center inner droplet
- Polymerize monomer in liquid 2



Criteria for Stable Droplet Centering

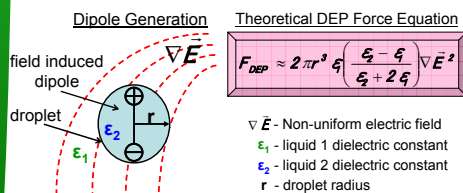
- Liquid 2 (outer droplet) must be immiscible in liquid 1 (ambient liquid) and liquid 3 (inner droplet)
 - Liquid 1 and 3 may be the same
- $\epsilon_2 > \epsilon_1$ (No constraint on ϵ_3)
- ρ_1, ρ_2, ρ_3 must be very similar
- Applied electric field must be uniform

Introduction to Dielectrophoresis

Dielectrophoresis - Response of a dielectric liquid, particle, or droplet to the force resulting from a non-uniform electric field (∇E).

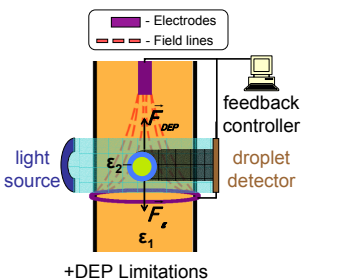
How DEP works:

- Field induces a dipole, which in turn generates a force along the field gradient
- Force direction depends on ϵ_1 and ϵ_2



+DEP Levitation Apparatus

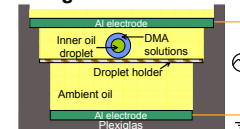
- Feedback controlled levitation
- Droplet moves to field maxima



- #### +DEP Limitations
- $\epsilon_2 > \epsilon_1$
 - Liquid 2 immiscible with liquid 1 and 3
 - Liquid 1 conductivity $< 10^{-4}$ S/m
 - Densities of liquid 1, 2, and 3 must be very similar
 - Monomer must dissolve only in liquid 2

Proof of Principle

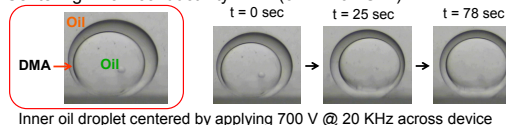
Centering device cross-section



Liquids and Densities
 Liquid 1 - Mixed silicone oils
 Liquid 2 - N,N-dimethylacetamide (DMA)
 Liquid 3 - Mixed silicon oils
 Densities slightly offset so $\rho_1 < \rho_2 < \rho_3$

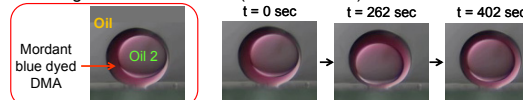
Successful Droplet Centering

Centering in low conductivity DMA ($\sigma = 2 \times 10^{-4}$ S/m):



Inner oil droplet centered by applying 700 V @ 20 KHz across device

Centering in conductive DMA ($\sigma = 1 \times 10^{-3}$ S/m):



Inner oil droplet centered by applying 700 V @ 20 KHz across device

Future Work

- Levitate double emulsions
- Test centering of different double emulsions
- Polymerize outer liquid in centered double emulsions
- Increase centering speed

Acknowledgments

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